Fisher Scientific

Material Safety Data Sheet
Hydrochloric acid 6N
MSDS\# 89586
Section 1 - Chemical Product and Company Identification

| MSDS Name: | Hydrochloric acid 6N |  |
| :--- | :--- | :--- |
| Catalog Numbers: | SA56-1, SA56-200, SA56-4, SA56-500 |  |
| Synonyms: | Muriatic acid; Chlorohydric acid; Hydrogen chloride solution. |  |
|  |  | Fisher Scientific |
| Company Identification: |  | One Reagent Lane |
|  | Fair Lawn, NJ 07410 |  |
| For information in the US, call: | $201-796-7100$ |  |
| Emergency Number US: | $201-796-7100$ |  |
| CHEMTREC Phone Number, US: | $800-424-9300$ |  |

Section 2 - Composition, Information on Ingredients

## Risk Phrases: 3437

| CAS\#: | $7647-01-0$ |
| :--- | :--- |
| Chemical Name: | Hydrogen chloride |
| $\%$ : | 22 |

EINECS\#: 231-595-7
Hazard Symbols:

Risk Phrases:

| CAS\#: | $7732-18-5$ |
| :--- | :--- |
| Chemical Name: | Water |
| $\%:$ | 78 |
| EINECS\#: | $231-791-2$ |

Hazard Symbols:

Text for R-phrases: see Section 16

Hazard Symbols:


Risk Phrases:

T C


2335

## Section 3 - Hazards Identification

EMERGENCY OVERVIEW
Danger! May be harmful if swallowed. May cause fetal effects based upon animal studies. Causes eye and skin burns. May cause severe respiratory tract irritation with possible burns. May cause severe digestive tract irritation with possible burns. Corrosive to metal. Target Organs: Respiratory system, teeth, eyes, skin.

## Potential Health Effects

May cause irreversible eye injury. Vapor or mist may cause irritation and severe burns. Contact with liquid is corrosive to the eyes and causes severe burns. May cause painful sensitization to light.

Skin: May be absorbed through the skin in harmful amounts. Contact with liquid is corrosive and causes severe burns and ulceration.

May cause circulatory system failure. Causes severe digestive tract burns with abdominal pain, vomiting, and possible death. May cause corrosion and permanent tissue destruction of the esophagus and digestive tract.
May cause severe irritation of the respiratory tract with sore throat, coughing, shortness of breath and delayed Inhalation: lung edema. Causes chemical burns to the respiratory tract. Exposure to the mist and vapor may erode exposed teeth. Causes corrosive action on the mucous membranes.

Prolonged or repeated skin contact may cause dermatitis. Repeated exposure may cause erosion of teeth. May Chronic: cause fetal effects. Laboratory experiments have resulted in mutagenic effects. Prolonged exposure may cause conjunctivitis, photosensitization, and possible blindness.

## Section 4 - First Aid Measures

Eyes:
Skin:
Get medical aid immediately. Do NOT allow victim to rub eyes or keep eyes closed. Extensive irrigation with water is required (at least 30 minutes). SPEEDY ACTION IS CRITICAL!
Get medical aid immediately. Immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Wash clothing before reuse. Destroy contaminated shoes.

Ingestion:

Inhalation:

Notes to
Physician:
Antidote:

General Information:

Extinguishing
Media:

Do not induce vomiting. If victim is conscious and alert, give 2-4 cupfuls of milk or water. Never give anything by mouth to an unconscious person. Get medical aid immediately. Give milk of magnesia.
Get medical aid immediately. Remove from exposure and move to fresh air immediately. If breathing is difficult, give oxygen. Do NOT use mouth-to-mouth resuscitation. If breathing has ceased apply artificial respiration using oxygen and a suitable mechanical device such as a bag and a mask.

Do NOT use sodium bicarbonate in an attempt to neutralize the acid.
Do NOT use oils or ointments in eye.

## Section 5 - Fire Fighting Measures

As in any fire, wear a self-contained breathing apparatus in pressure-demand, MSHA/NIOSH (approved or equivalent), and full protective gear. Water runoff can cause environmental damage. Dike and collect water used to fight fire. During a fire, irritating and highly toxic gases may be generated by thermal decomposition or combustion. Not flammable, but reacts with most metals to form flammable hydrogen gas. Use water spray to keep fire-exposed containers cool. Vapors may be heavier than air. They can spread along the ground and collect in low or confined areas. Reaction with water may generate much heat which will increase the concentration of fumes in the air. Containers may explode when heated.
For large fires, use water spray, fog, or alcohol-resistant foam. Substance is nonflammable; use agent most appropriate to extinguish surrounding fire. Do NOT get water inside containers. Do NOT use straight streams of water. Most foams will react with the material and release corrosive/toxic gases. Cool containers with flooding quantities of water until well after fire is out. For small fires, use carbon dioxide (except for cyanides), dry chemical, dry sand, and alcohol-resistant foam.
Autoignition
Temperature:
Flash Point: Not applicable.

## Explosion Not available <br> Limits: Lower:

Explosion
Limits: Upper:
NFPA Rating: health: 3; flammability: 0 ; instability: 1 ;

## Section 6 - Accidental Release Measures

General Information:

Use proper personal protective equipment as indicated in Section 8.
Large spills may be neutralized with dilute alkaline solutions of soda ash (sodium carbonate, Na 2 CO 3 ), or lime (calcium oxide, CaO ). Avoid runoff into storm sewers and ditches which lead to waterways. Clean up

Spills/Leaks: spills immediately, observing precautions in the Protective Equipment section. Remove all sources of ignition. Provide ventilation. Do not get water inside containers. A vapor suppressing foam may be used to reduce vapors. Cover with dry earth, dry sand, or other non-combustible material followed with plastic sheet to minimize spreading and contact with water.

Section 7 - Handling and Storage
Wash thoroughly after handling. Remove contaminated clothing and wash before reuse. Use only in a wellventilated area. Contents may develop pressure upon prolonged storage. Do not breathe dust, mist, or vapor. Do

Handling:
not get in eyes, on skin, or on clothing. Keep container tightly closed. Do not ingest or inhale. Discard contaminated shoes. Use caution when opening. Keep from contact with moist air and steam.
Do not store in direct sunlight. Store in a tightly closed container. Store in a cool, dry, well-ventilated area away Storage: from incompatible substances. Corrosives area. Do not store in metal containers. Do not store near flammable or oxidizing substances (especially nitric acid or chlorates).

Section 8 - Exposure Controls, Personal Protection


OSHA Vacated PELs: Hydrogen chloride: None listed Water: None listed

## Engineering Controls:

Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower. Use adequate general or local exhaust ventilation to keep airborne concentrations below the permissible exposure limits.
Exposure Limits
Personal Protective Equipment
Eyes: Wear chemical splash goggles and face shield.
Skin: Wear neoprene or polyvinyl chloride gloves to prevent exposure.
Clothing: Wear appropriate protective clothing to prevent skin exposure.
Respirators: A respiratory protection program that meets OSHA's 29 CFR 1910.134 and ANSI Z88.2 requirements or European Standard EN 149 must be followed whenever workplace conditions warrant respirator use.

## Section 9 - Physical and Chemical Properties

> Physical State: Clear liquid
> Color: colorless to slight yellow
> Odor: strong, pungent
> pH: 0.01
> Vapor Pressure: $5.7 \mathrm{~mm} \mathrm{Hg} @ 0$ deg C
> Vapor Density: 1.26
> Evaporation Rate: > 1.00 (N-butyl acetate)
> Viscosity: Not available
> Boiling Point: $81.5-110$ deg C @ 760 mmHg
> Freezing/Melting Point: -74 deg C $\left(-101.20^{\circ} \mathrm{F}\right)$
> Decomposition Temperature: Not available
> Solubility in water: Miscible
> Specific Gravity/Density: $1.0-1.2$
> Molecular Formula: $\mathrm{HCl} . \mathrm{H} 2 \mathrm{O}$
> Molecular Weight: 36.46
> Section $10-$ Stability and Reactivity

Chemical Stability: Stable under normal temperatures and pressures.
Conditions to
Avoid:
Mechanical shock, incompatible materials, metals, excess heat, exposure to moist air or water, bases.

Incompatibilities
with Other
Materials

Bases, acetic anhydride, alkali metals, aluminum, amines, copper, copper alloys, fluorine, iron, sodium hydroxide, steel, sulfuric acid, vinyl acetate, zinc, potassium permanganate, cesium acetylene carbide, rubidium acetylene carbide, rubidium carbide, sodium, chlorosulfonic acid, oleum, carbonates, perchloric acid, calcium phosphide, metal oxides, acetates, cesium carbide, beta-propiolactone, ethyleneimine, propylene oxide, lithium silicides, alcohols + hydrogen cyanide, 2-aminoethanol, ammonium hydroxide, calcium carbide, 1,1-difluoroethylene, ethylene diamine, magnesium boride,
mercuric sulfate, uranium phosphide.
Hazardous
Decomposition Hydrogen chloride, chlorine, carbon monoxide, carbon dioxide, hydrogen gas.
Products
Hazardous
Polymerization

|  | Section 11 - Toxicological Information |
| :---: | :---: |
| RTECS\#: | CAS\# 7647-01-0: MW4025000 MW4031000 CAS\# 7732-18-5: ZC0110000 |
|  | RTECS: <br> CAS\# 7647-01-0: Inhalation, mouse: LC50 $=1108 \mathrm{ppm} / 1 \mathrm{H}$; Inhalation, mouse: LC50 $=20487 \mathrm{mg} / \mathrm{m} 3 / 5 \mathrm{M}$; Inhalation, mouse: LC50 $=3940 \mathrm{mg} / \mathrm{m} 3 / 30 \mathrm{M}$; Inhalation, mouse: LC50 $=8300 \mathrm{mg} / \mathrm{m} 3 / 30 \mathrm{M}$; Inhalation, rat: LC50 $=3124 \mathrm{ppm} / 1 \mathrm{H}$; |
| LD50/LC50: | Inhalation, rat: LC50 $=7004 \mathrm{mg} / \mathrm{m} 3 / 30 \mathrm{M}$; <br> Inhalation, rat: LC50 $=45000 \mathrm{mg} / \mathrm{m} 3 / 5 \mathrm{M}$; <br> Inhalation, rat: LC50 $=8300 \mathrm{mg} / \mathrm{m} 3 / 30 \mathrm{M}$; <br> Oral, rabbit: LD50 $=900 \mathrm{mg} / \mathrm{kg}$; |
|  | RTECS: <br> CAS\# 7732-18-5: Oral, rat: LD50 $=>90 \mathrm{~mL} / \mathrm{kg}$; |
| Carcinogenicity: | Hydrogen chloride - IARC: Group 3 (not classifiable) <br> Water - Not listed as a carcinogen by ACGIH, IARC, NTP, or CA Prop 65. |
| Other: | Rinsed with water test: Administration into the eye (rabbit) $=5 \mathrm{mg} / 30 \mathrm{sec}($ Mild $)$. Section 12 - Ecological Information |
| Ecotoxicity: | Fish: Bluegill/Sunfish: $3.6 \mathrm{mg} / \mathrm{L} ; 48 \mathrm{Hr}$, Lethal (unspecified) Fish: Bluegill/Sunfish: LC50; 96 Hr ; $\mathrm{pH} 3.0-3.5$ |

Section 13 - Disposal Considerations
Dispose of in a manner consistent with federal, state, and local regulations.
Section 14 - Transport Information
US DOT
Shipping Name: HYDROCHLORIC ACID
Hazard Class: 8
UN Number: UN1789
Packing Group: II
Canada TDG
Shipping Name: Not available
Hazard Class:
UN Number:
Packing Group:

USA RQ: CAS\# 7647-01-0: 5000 lb final RQ; 2270 kg final RQ
Section 15 - Regulatory Information
European/International Regulations
European Labeling in Accordance with EC Directives
Hazard Symbols: T C
Risk Phrases:
R 23 Toxic by inhalation.
R 35 Causes severe burns.
Safety Phrases:

S 1/2 Keep locked up and out of reach of children.
S 9 Keep container in a well-ventilated place.
S 26 In case of contact with eyes, rinse immediately with plenty of water and seek medical advice.
S 36/37/39 Wear suitable protective clothing, gloves and eye/face protection.
S 45 In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible).

WGK (Water Danger/Protection)
CAS\# 7647-01-0: 1
CAS\# 7732-18-5: Not available

## Canada

CAS\# 7647-01-0 is listed on Canada's DSL List
CAS\# 7732-18-5 is listed on Canada's DSL List
Canadian WHMIS Classifications: E, D2A
This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations and the MSDS contains all of the information required by those regulations.
CAS\# 7647-01-0 is listed on Canada's Ingredient Disclosure List
CAS\# 7732-18-5 is not listed on Canada's Ingredient Disclosure List.

## US Federal

TSCA
CAS\# 7647-01-0 is listed on the TSCA
Inventory.
CAS\# 7732-18-5 is listed on the TSCA
Inventory.
Section 16 - Other Information
MSDS Creation Date: 7/06/1999
Revision \#7 Date 7/20/2009

The information above is believed to be accurate and represents the best information currently available to us. However, we make no warranty of merchantibility or any other warranty, express or implied, with respect to such information, and we assume no liability resulting from its use. Users should make their own investigations to determine the suitability of the information for their particular purposes. In no event shall the company be liable for any claims, losses, or damages of any third party or for lost profits or any special, indirect, incidental, consequential, or exemplary damages howsoever arising, even if the company has been advised of the possibility of such damages.

